

Novel Heterocyclic Compounds for Cancer Chemotherapy

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Abstract— *Heterocyclic molecules are very important to medical chemistry because they are used to make drugs, especially chemotherapy. The goal of this study is to build on existing manufacturing methods for heterocyclic drugs so that they can be used more effectively in cancer medicine. We want to find new ways to make compounds and get around problems that come up with making complex heterocyclic structures. This will help us find chemicals that are very good at fighting cancer. Biocatalysis and flow chemistry are two current synthetic techniques that we use in our method to speed up the synthesis process, protect the environment, and increase yield and purity. As an environmentally friendly option to standard chemical methods, biocatalysis uses the precision and efficiency of enzymes to speed up processes in mild conditions. Flow chemistry, on the other hand, lets synthesis happen all the time, which makes it easier to direct reactions and make them bigger. For our study, we made a bunch of new heterocyclic chemicals and tested how well they killed different kinds of cancer cells. In early tests, a number of chemicals have shown promise in fighting cancer, showing that they could be used as treatment drugs. Structure-activity relationship (SAR) studies have been done to figure out what about the molecules makes them work, which will help make these chemicals even better. We also used computer chemistry tools to guess how these heterocyclic molecules would react with proteins that play a role in the growth of cancer. These predictions are confirmed by tests done in vitro and in vivo, which give us a full picture of how the drug works and its possible uses in therapy.*

Keywords— *Cancer chemotherapy, cancer stem cells mechanisms of action, side effects.*

I. INTRODUCTION

Cancer is the second most common cause of death in people, after heart disease. These days, early diagnosis and the right care are helping hundreds of thousands of people with cancer live longer. Most of the cells in our bodies are specialized, which means they have a shape and set of traits that are unique to the job they do. Normal cells and differentiated cells grow together in a single, well-organized layer under the control of controlled processes like contact inhibition. Most cancer cells can divide quickly, not differentiate properly, invade nearby tissue, and start new growth in places they shouldn't be. These are the main differences between normal cells and most cancer cells.

Normally, cells can only go through the cell cycle about fifty times before they die. But cancer cells can go through and out of the cycle infinite times. Most cancer cells have nuclei that are large and have a lot of chromosomes. When cancer cells divide, they form tumors, which are abnormal groups of cells that invade and kill nearby tissues. It looks like an encapsulated mass that is disordered but doesn't go through nearby tissue. This type of tumor is benign. The second type of tumor is one that has gotten out of hand. These tumors have abnormal, uncontrollable cell growth along with a loss of organization in some ways. At different times during the illness, malignant tumors invade nearby organs.

Cancer is a genetic disease that is usually caused by things in the environment. Carcinogens are chemicals that can be found in a lot of popular foods, drinks, air, and outdoor factors, like sunlight. A mutation in a single normal cell is where most cancers begin. A mutagen is any chemical that can change the DNA code; these chemicals are also called toxins. But mistakes made by DNA polymerase while DNA is being copied can also cause changes. Bishop et al. (1987) say that cancer can show up in many different ways, affecting many different organs and tissues and even growing in many different ways within the same tissue. Cell division is the process by which normally growing cells turn into cells that differentiate wrongly. This is its beginning. In the second stage, cancer cells metastasize, which makes it hard to treat a single cell that has spread to another

part of the body. Another study by Kundsonet. et. al. (2010) looked at how molecular processes inside and outside of cells can control cell growth and division. Changes in genes that control development, virus diseases, and higher levels of growth hormones that stimulate cells can all make editing impossible.

Cancer is a group of difficult diseases that starts with cells dividing in a way that isn't normal and can spread to other parts of the body. When cells don't differentiate properly, people often lose weight, cough up a lot of mucus, and get lumps that don't belong. Cancer rates are going up all over the world. Cancer is expected to cause 19.3 million new cases and almost a million deaths each year by 2020. It is the second most common cause of death in the world. Based on these numbers, just over half of the new cases that happen each year end in death. Some types of cancer have been found in the esophagus, breast, cervix, and colon areas, but esophageal, breast, and lung cancers are the most common. Prostate cancer is most common in older men. They found that age and sex have a big effect on the chance of getting cancer and the types of treatments that can be used. Some research suggests that men are more likely to get illnesses than women.

Vascular networks help cancer cells get air, nutrients, and waste products, all of which are necessary for their survival and growth from the start. Angiogenesis is the process by which new blood vessels are made. When there is no arterial extension, cancer cells die or go through death, which slows the growth rate. The amount of factors that cause cells to die, or necrosis, is a sign of how severe a tumor is. Usually, the process of making new blood vessels starts when the basement membrane breaks, letting out angiogenic factors. When vascular cells are told to move, multiply, and stay stable, a shield of protection is made. Controlling tumor vascularization has been the focus of many experts who want to slow the development of cancer. It is important to carefully target the activators of arterial growth and know how they control this chemical signal in order to keep things in check. They will not get enough food because of this.

II. LITERATURE REVIEW

Bhawal Ganesh Shivaji (2021) A major worldwide health problem, the rise of antibiotic-resistant bacteria calls for the search for new and improved antimicrobials. Various heterocyclic compounds have different chemical structures and pharmacological effects, which makes them a promising source for new antibacterial medicines. To that end, this review article will survey the literature on heterocyclic compound production and antimicrobial activity testing. We will talk about several synthetic techniques and methods for producing heterocyclic scaffolds, and how they work against bacterial infections. In addition, we will take a close look at the obstacles and present research on heterocyclic compounds as antibacterial agents.

Sharma (2023) Worldwide, cancer is responsible for the deaths of millions of people. It is very necessary to develop novel anticancer medications. because to the inadequacy and side effects of current chemotherapy. Among the most significant chemical frameworks demonstrating anticancer action is the thiazolidin-4-one scaffold. Extensive study on thiazolidin-4-one derivatives has indicated that these chemicals have strong anticancer effects, according to recent scholarly articles. This review focusses on thiazolidin-4-ones and its anticancer effects via the inhibition of different enzymes and cell types. It also discusses the several ways these compounds may be synthesized, including synthetic, green, and nanomaterial-based methods. Scientists may find this article's comprehensive overview of current best practices in the area to be both intriguing and useful as they investigate these heterocyclic compounds further for potential anticancer effects.

Rao (2021) Eighty percent of commercial pharmaceuticals use heterocyclic moieties as their fundamental skeleton, based on the 2014–2015 US retail market. But a lot of artificial processes aren't long-lasting, therefore we need tactics that are less harmful to the environment. As an example, molecules may be quickly and efficiently synthesized with high yields utilizing minimal energy by microwave-assisted synthesis. Moreover, the use of metal-impregnated nanoparticles offers many benefits in nanoparticle-catalyzed synthesis, such as the capacity to recycle the catalyst, achieve high yields, and accelerate reaction times. Additional ecologically friendly methodologies include Water-based organic synthesis, solvent-free synthesis, combinatorial synthesis, and sonochemical synthesis, and synthesis facilitated by ionic liquids. We examine the synergistic relationship between organic synthesis, solvent-free synthesis, microwave radiation, and organic synthesis in water. The application of nanoparticles is discussed below. As catalysts in the production of complicated heterocyclic compounds. We highlight environmentally friendly features of synthetic processes.

Javahershenas, Ramin. (2022). Lately, the process of creating heterocyclic compounds has been has garnered significant interest from organic and medicinal chemists, who have investigated a wide range of materials. As an effective, affordable, adaptable, and versatile intermediate, phenacyl bromide is one of numerous organic molecules that may be synthesized in various chemical processes. This article provides a synopsis of phenacyl bromide's important uses, with an emphasis on its function in multicomponent reactions and its involvement in recent synthetic breakthroughs up to the end of 2021.

Sharma, Shivali & Utreja, Divya. (2021). Because of their ubiquitous presence in nature and their many useful uses in many different areas, including medicine, agriculture, photochemistry, biocidal formulations, polymer science, and substantial clinical usage, heterocyclic moieties constitute a significant portion of organic chemistry. They have medicinal qualities that make them a potential weapon against several infectious illnesses. Virus infections are among the most frequent infectious illnesses, and they pose a significant threat to public health throughout the globe. In order to extend people's lives, it is critical to find and develop antiviral medications and therapeutic approaches that can ward against different types of viruses. To assist researchers and other stakeholders better understand the topic at hand, this study provides a synopsis of all heterocyclic compounds synthesized and tested for antiviral activity from 2015 forward. In an effort to discover novel, potentially effective antiviral medicines, many alterations were considered centred on the various heterocyclic scaffolds.

III. TRANSFORMING CHEMICALS

Heterocyclic compounds are not like homocyclic compounds, which only have carbon atoms in their rings. Heterocyclic compounds have rings made up of different types of atoms. When you look at molecules, an organic heterocyclic substance is one that has at least one carbon atom in each of its rings. Ring atoms that don't have carbon are all hetero atoms. Different types of atoms are made up of the three most common elements: sulfur, nitrogen, and oxygen. A huge number of chemicals with heterocyclic rings are useful in environmental engineering and study, and a lot of these molecules are also very important for living things. A hetero atom ring can be made out of any element except for alkali metals, according to the idea. It's not just the type of ring atoms that matter; the number of them also shows how big the ring is. Still, rings with three members are the smallest. Rings with five or six members are the smallest, and heterocyclic rings are the most important. Triazoles, thiadiazole, and thiazolidinone are all chemical molecules that have five parts that are not in the same ring. These chemicals can help with many biological processes, such as controlling seizures, killing fungi and bacteria, reducing inflammation, protecting cells from damage, and lowering histamine levels. They are very important in medical chemistry.

In their ring shape, most heterocyclic molecules have carbon plus at least one element, such as sulfur, oxygen, or nitrogen. They can also be artificial. Most people think that non-carbon atoms have replaced carbon atoms, which is why they are often called heteroatoms. They could be made up of aromatic or non-aromatic rings. Heterocyclic chemistry is the study of heterocycles, including how they are made, what they are like, and what they might be used for. Heterocyclic derivatives can be broken down into two main groups: aromatic and non-aromatic. The aromatic product furan is shown by the first row of rings in Figure 1, which has five members.

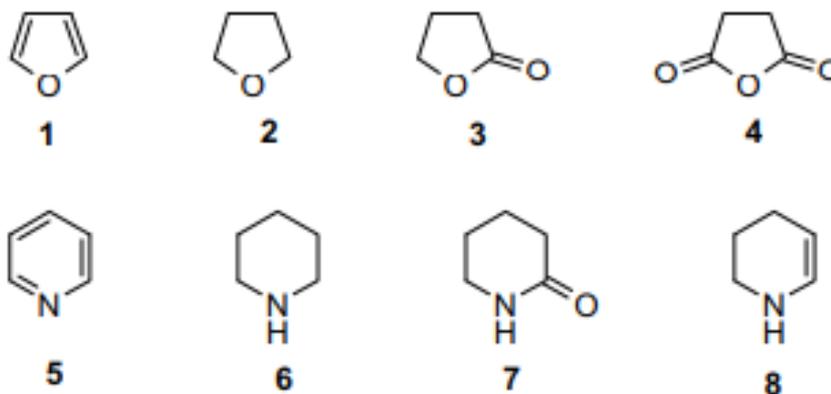
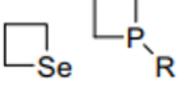


FIGURE 1: Heterocyclic compounds, with examples provided

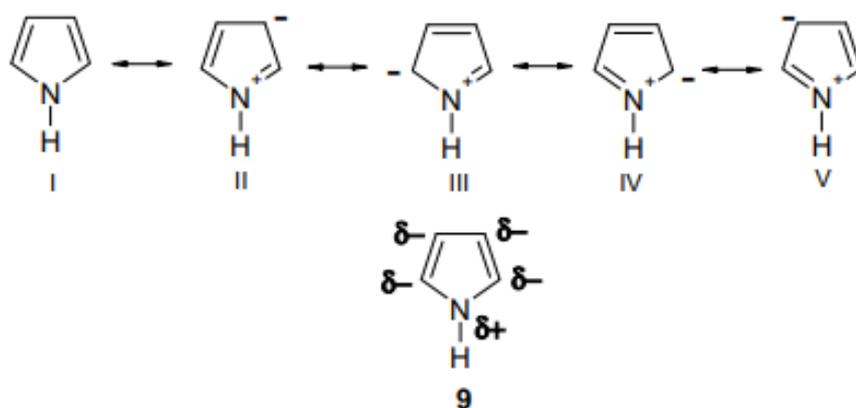
In the second row, the six-membered rings are aromatic at first (pyridine (5)), but they lose their aromaticity as you move down the list. These are piperidin-2-one (7), piperidine (6), 1, 2, 3, 4-tetrahydropyridine (8), and more. They react like an amine, an enamine, or an amide, based on the order in which they are present. When you mix the expected reactivity of aromatic systems with heteroatoms, you get aromatic heterocycles. These have a more complicated reactivity profile than non-aromatic systems, which are otherwise very similar to normal non-cyclic forms. Because of this, most books on heterocyclic chemistry are mostly about how aromatic molecules respond. The tables 1.1–1.4 in these books show heterocyclic derivative models. In Table 1.1, you can see simple heterocyclic systems with three or four members. Their expected reaction is always linked to the ring strain, which gives off energy when it comes in contact with aliphatic chemicals.

TABLE 1
PRIMARY HETEROCYCLES WITH THREE OR FOUR MEMBERS

Ring Size	Heteroatom			
	N	O	S	Other
3	 Aziridine	 Oxirane	 Thiirane	
	 Diaziridine	 Dioxirane		 Oxaziridine
4	 Azetidine	 Oxetane	 Thietane	 Seletane Phosphetane

3.1 Characteristics and Responses of Aromatic Five-Membered Compounds:

The reaction time of an individual to prominent aromatic heterocycles is often elucidated using a graphical valence bond resonance framework, as shown in most educational resources on heterocyclic chemistry. We examine two examples that exemplify the majority of aromatic rings: pyrrole, which symbolizes p-excessive rings, and pyridine, which symbolizes p-deficient rings. Pyrrole demonstrates electrical neutrality in its isoelectronic link with the cyclopentadienyl anion; its aromatic sextet comprises a nitrogen atom with two electrons; and the sound it produces. A hybrid may be represented as a combination of kinds I through V (Scheme 1.1), with one type being entirely devoid of charges while the others exhibit separated charges. Diverse types, as anticipated, influence the structure of pyrrole in distinct ways. The charged and uncharged forms of nitrogen, which use their own pair of electrons, are particularly relevant. Forms III, IV, II, and V are shown here. Structure 9, as a whole, suggests that the electrical density of the carbon sites exceeds that of the typical aromatic system, benzene, and the heteroatom has a partial positive charge. Consequently, electrophiles, as opposed to nucleophiles, would easily engage with a π -excessive system such as pyrrole.



SCHEME 2: A variety of pyrrole resonance hybrids

IV. CHEMOTHERAPY

Chemotherapy is defined as the use of chemical agents to treat disease. Chemotherapeutic agents are chemical compounds used. The paramount attribute of efficacious chemotherapeutic agents is their pronounced toxicity selectivity for a particular germ; this allows for the inhibition or eradication of bacteria throughout the body at levels that do not harm host cells. Tumor cell resistance to chemotherapeutic agents is a significant challenge in cancer treatment; thus, a comprehensive arsenal of selective and potent compounds is required to tackle cancer-associated proliferation difficulties. Anticancer medicines are classified into many groups based on their mechanisms of action (refer to figure 2). The majority of chemotherapeutic agents may induce the death of tumor cells, either directly or indirectly.

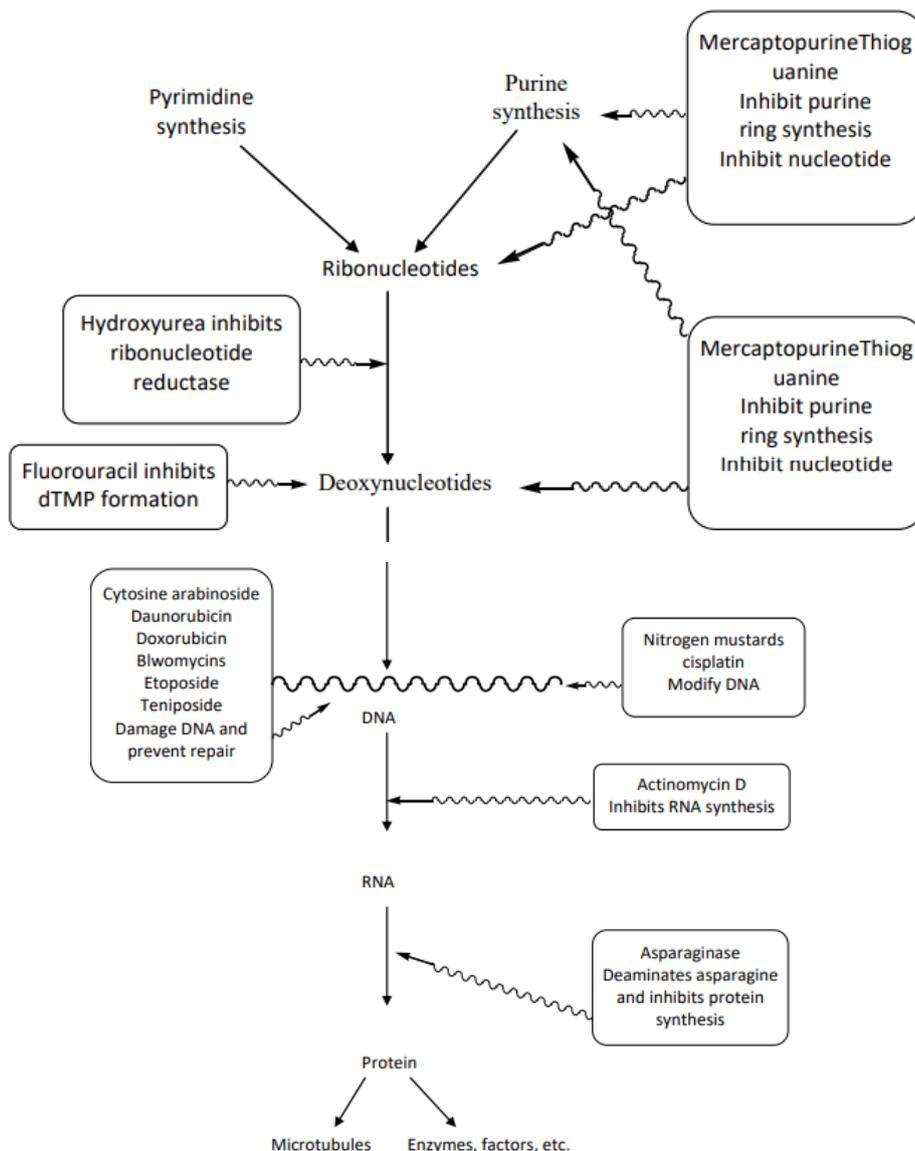
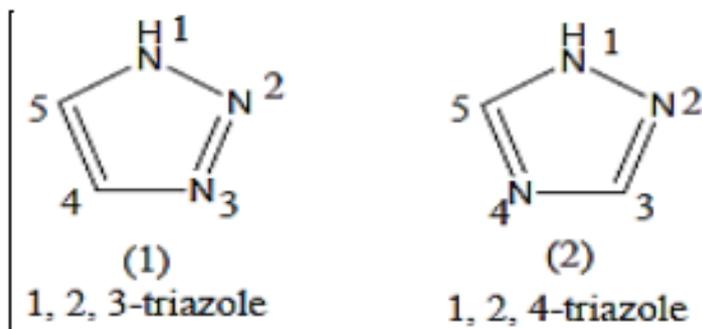


FIGURE 2: Potential cancer-fighting medication sites of action

4.1 Triazole:

There are three nitrogen and three carbon atoms in a triazole, making it a heterocyclic molecule with five members. One kind of triazole contains a nitrogen that is similar to pyrrole while the other has two nitrogen’s that are similar to pyridine. In early 1885, Bladin identified variants of the carbon nitrogen ring system C₂N₃H₃, and the term "triazole" was first used to them, even though the structure was somewhat off³⁵. There is a chance of tautomerism in 1, 2, and 3-triazole in both classes of triazoles, and these tautomers are essentially the same.



4.2 Types of Chemotherapy:

Many distinct chemotherapies exist for the treatment of cancer; these chemo medicines are categorized according to their chemical structures and the way they interact with cancer cells. As new medications are developed, these categories might be changed. Some categorization is helpful for understanding the action processes, even if some medications function with distinct groups.

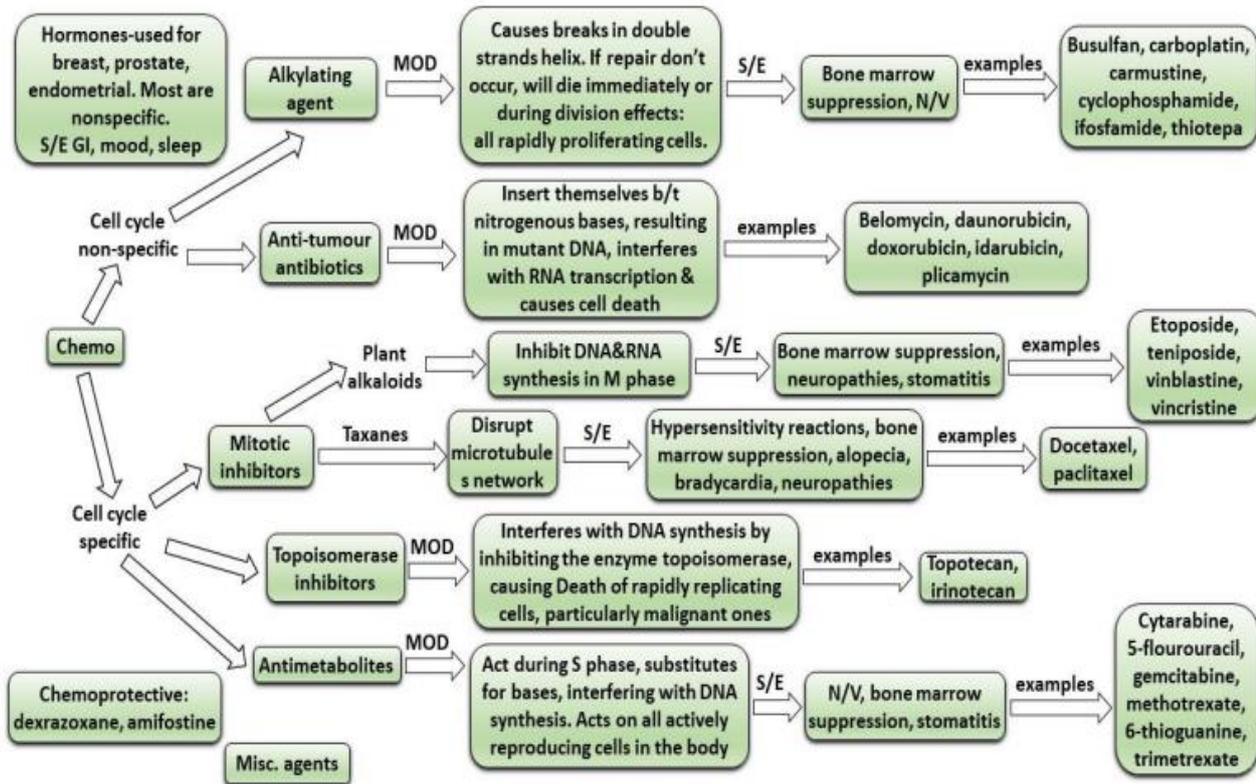


FIGURE 3: Chemotherapy medication categorisation according to mode of effect

4.3 Cancer Therapy:

Cancer risk factors, encompassing both hereditary and environmental components, are responsible for the majority of cancer cases. However, many of these factors are within our control or can be mitigated. The primary objective of mitigating cancer risk factors through modifications in lifestyle and environmental conditions is a crucial strategy for decreasing the cancer burden. Sexually transmitted infections, obesity, alcohol consumption, and tobacco exposure are risk factors that could potentially prevent nearly 30% of cancer-related deaths.

Early and reliable diagnosis of premalignant stages of tumor growth may facilitate timely intervention, representing an effective treatment option. Early detection of cancer, prior to metastasis, frequently allows for curative treatment options. In the initial, premalignant phases of colon cancer, minor surgical interventions may completely eliminate the disease. A variety of cancer treatment options exist; however, the selection of a specific treatment depends on the type of cancer, its location, and the stage of the disease. Cancer treatment has traditionally focused on three primary modalities: systemic therapy, radiation therapy, and surgical intervention. Systemic treatment, surgery, and radiation therapy represent prevalent combinations for cancer management.

Hormonal treatment, immunotherapy, chemotherapy, and targeted medications represent various forms of systemic treatments. The hormone-dependent characteristics of breast and prostate cancers, which develop or progress in reaction to hormonal influences, render hormone therapy a critical element of their treatment strategies. Targeted treatments are essential for certain cancer types as they inhibit disease progression and dissemination by disrupting specific molecules involved in these processes. Cancer cells exhibit heightened susceptibility to chemotherapy due to their rapid proliferation. This treatment functions by inhibiting the cell cycle, thereby diminishing or eradicating the capacity of cancer cells to replicate. Anticancer medications can be categorized as stage-specific, which target cancer cells during specific phases of proliferation or rest, or non-specific, which affect cancer cells across the entire cell cycle. Chemotherapy can be rendered more selective by targeting cancer cells

rather than healthy cells. All cells, including cancer cells, are surrounded by a dense layer of sugar-containing molecules known as polysaccharides. This category of polysaccharides demonstrates structural diversity across various tissues and organs in the body.

The chemical compositions of polysaccharides in cancerous regions differ from those in normal tissue. Consequently, the adverse effects of chemotherapy may be markedly diminished through the use of an appropriate drug carrier capable of identifying specific polysaccharides (Longmuir et al., 2009). Some cancers may be treated more effectively with a combination of chemotherapy agents. Effective combined chemotherapy treatments must target tumors at multiple levels and address cancer cells through distinct pathways. The first medication may inhibit DNA replication, whereas the second may obstruct protein synthesis.

V. CONCLUSION

Research into novel synthetic strategies for heterocyclic compounds has produced notable advancements in organic synthesis methodology and the potential for new chemotherapy treatments. This study effectively addresses key aspects of synthetic chemistry and its pharmaceutical applications, resulting in promising outcomes for cancer treatment. The incorporation of contemporary synthetic methodologies, including biocatalysis and flow chemistry, has shown significant advancements in the synthesis of heterocyclic compounds. These methods demonstrate reductions in reaction times, enhancements in yields, and a decrease in the utilization of hazardous reagents, thus fostering more sustainable and efficient synthetic processes. The synthesis and screening of diverse heterocyclic compounds have identified several candidates with notable cytotoxicity against various cancer cell lines. The findings indicate the potential of these novel compounds as effective agents in chemotherapy, presenting new opportunities for cancer treatment. Comprehensive SAR studies have clarified the molecular characteristics that enhance the anticancer efficacy of the synthesized compounds. This comprehension has informed the optimization of lead compounds, improving their efficacy and specificity toward cancer cells. Computational chemistry tools have yielded significant predictions about the interactions between heterocyclic compounds and target proteins associated with cancer proliferation. The predictions have been validated via *in vitro* and *in vivo* experiments, providing a thorough understanding of the mechanisms of action of these compounds. Optimizing synthetic routes for scalability and environmental sustainability facilitates the production of promising compounds in large quantities for subsequent development and clinical trials. The scalability is essential for moving from laboratory research to practical pharmaceutical application.

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